City College

B.Sc. Part III (1+1+1) Chemistry Honours (CEMA) Practical Examination-2020

Paper VIIIA (ORGANIC QUALITATIVE ANALYSIS AND LAB QUIZ)

Full Marks: 50

- Given compound's name : 4-Nitro benzoic acid,
 Write about the following
 - (i) Solubility and solubility classification
 - (ii) Lassaigne's test
 - (iii) Detection of functional groups(a)Nitrogenous and (b) Non-nitrogenous

$$(\frac{1}{2} \times 4 + 2) + (2 \times 3) + (2\frac{1}{2} \times 6) = 25$$

- 2. Lab quiz 2 x5=10
- (i) Sometimes we get a black precipitate on addition of ferrous sulphate to the sodium extract in Lassaigne's test—Why? Explain it by equation.
- (ii) p-Toluidine offers a precipitate when treated with Brady's reagent, although it does not contain a carbonyl group. Explain it with chemical reactions.
- (iii) Which of the following compound/compounds will not give Lassaigne's test for nitrogen and why?

$$\begin{array}{ccc}
\bigoplus \\
NH_2-NH_3 \text{ HSO}_4^{\Theta} & PhNHNH_2 & NH_2CONH_2 \\
I & II & III
\end{array}$$

(iv) Which of the following compound can respond to dye test and why?

- (v) What happens when phenol is treated with excess Br₂-water, write down the structure of the product formed.
- 3. Internal assessment

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Paper VIB (ORGANIC SPECTRA)

Full Marks: 25

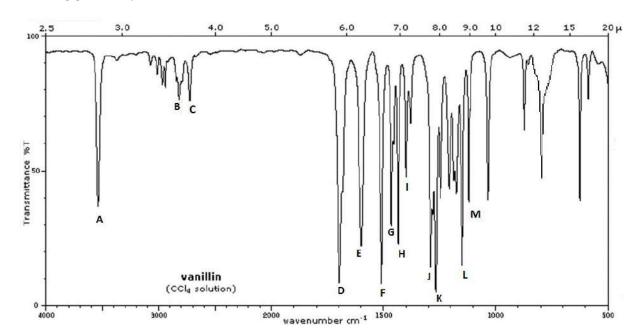
1. Identify structure of the given molecule showing different type of hydrogen.

2

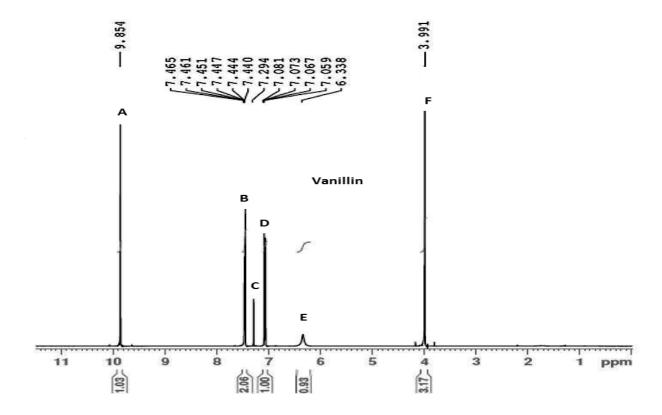
Name of the molecule: Vanillin

2. Using given IR spectra

6



Frequency (cm ⁻¹)	Identification of Bands in the Spectrum	Assignment of Bands	Nature of Bands
	(½ x 4)	(½ x 4)	(½ x 4)
3550			
2750			
1700			
1600			



Signals	Identification of Signals	Splitting of	No. of H atoms	Assignment of	Explanation
	1x3	Signals	1/2 x 3 = 1½M	Signals	1x3
		1/2 x 3 = 1½M		1x3	
9.85					
6.33					
3.99					

4. Internal assessment 5